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## Thomas A. Edison's Latest Invention

## A Storage Battery Designed and Constructed from the Automobile User's Point of View

By Joseph B. Baker, E.E.

For over a decade, Thomas A. Edison has been working to produce a type of storage battery better than the old lead type. At last he has succeeded. His new cells are now in actual commercial use. In the following article the reader will find a description of the new battery, written after a painstaking study of its manufacture.



LD, indeed, is the idea of the commercial storage battery. Lead plates were immersed in an acid electrolyte, by Gaston Planté in 1861. Emile Faure brought out the pasted-plate battery in 1879, and Charles F. Brush in 1881 introduced improvements which made

the lead battery a factor in electric vehicle work, producing a cell which would give eight watts of

electric power per pound of material. From these early beginnings, many inventors have sought to increase the electrical efficiency, and the durability or "life" of the cell, and reduce the cost of manufacture.

It was in the early eighties that the public began to think about "stored electricity" which could be bought by the can, as it were, like any other commodity. This dream has come true in the new Edison nickel-iron storage cell. Stored electricity finds its greatest usefulness in propelling cars and road vehicles, and it has been for this application, primarily, that the Edison storage battery has been de-

veloped. The need of the electric vehicle was recognized by Mr. Edison nine years ago. He saw that there are two viewpoints: that of the electrical man with his instruments, his rules for efficient operation and reasonable life of the battery, his absolute knowledge that the same care should be given a vehicle battery that is given a valued horse or even a railroad locomotive; and that of the automobile driver, who simply wishes to go somewhere with his car, and who, when he arrives somewhere, wishes to go back. And in this longpromised battery the highly practical nature of Mr. Edison's work is once more exemplified in that he has held uncompromisingly to the automobilist's point of view. The fact to be

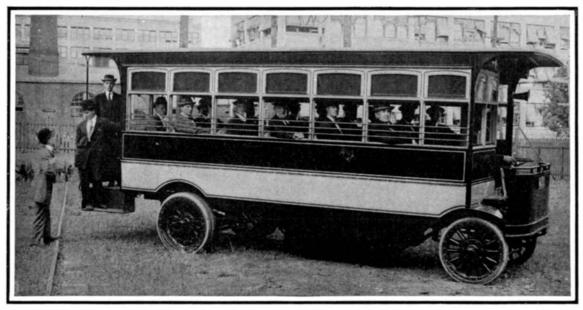
faced, he realized, is that the electric vehicle user will work his car to the utmost, and will be limited only by repairs and breakdowns: just as the average gasoline car driver has always operated his car to the limit of repairs and breakdowns. How well he has succeeded in aligning his battery to this ideal, the automobile driver will appreciate from the figures of actual performance that are now available. A recent average of 144.35 miles on a single charge was made in a family-type electric vehicle equipped with the new battery. This figure is the average of four trips, each with New York city (Manhattan) as the starting point. The longest of these trips, 172.1

miles, was to Babylon, Long Island, by way of Mineola and Plainview, returning by the southerly route through Massapequa and Freeport. The map distance by this route is 90.9 miles, and 81.2 miles additional was covered after the regular run, the average speed being 10.93 miles per hour.

The original Edison storage battery, known as the type E, was put out about seven years ago. The type E cell marked a definite step in invention, in that it proved the commercial success of a wholly new voltaic combination, a radical departure from the lead plates in an acid electrolyte. Believing that the lead cell had too many inherent weaknesses to realize the

full promise of "stored electricity" as a motive power for vehicles, Mr. Edison had set his staff to work on an entirely new 'line, and the outcome was a nickel-iron element immersed in an alkaline electrolyte, lighter and cleaner than the lead cell, with lower cost of operation and upkeep to offset its higher initial cost, and possessing hitherto unheard-of properties of remaining undeteriorated either by overcharging or being left uncharged. These were the very qualities demanded by the automobile user. The cell developed certain weaknesses in service. which caused the inventor to withdraw it from the

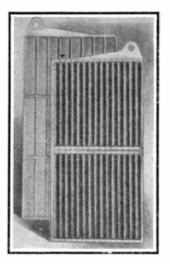
market and resume experiments to improve the form of the voltaic combination, in order to obtain better electrical properties and a longer life. The new series of experiments lasted four years, and resulted in the present commercial Edison cell, known as the type A. This cell is now being actively marketed for electric vehicle work in three sizes, the A-4, A-6, and A-8 cells, having respectively 4, 6, and 8 positive plates and 5, 7, and 9 negative plates, the two outside plates of the element being negatives. The normal discharge voltage is 1.2, and a battery is regarded as completely discharged when it shows an average of 1.0 volt per cell. In each size the dimensions of the individual (Continued on page 45.)



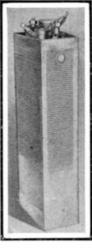
A storage battery omnibus. Mr. Edison is in the front seat.



Interior of type 4-A Edison cell and cover of can with terminal stuffing box, gas valve, and filling aperture.

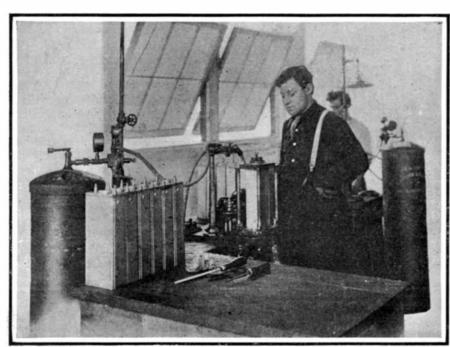


Positive plate at the rightwith nickel-hydrate active material in perforated tubes. Negative plate at the left with iron oxide active material in perforated flat pockets.

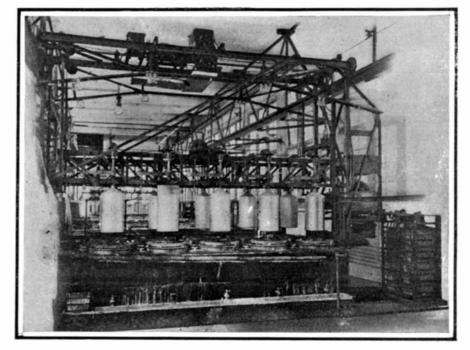


Element removed from can of Edison type A-4 cell. This cell is not in actual automobile use.





Welding the steel cans for the cell.



Machine for electro-deposition of nickel for use in positive plates.

(Continued from page 30.) of the containing jar or "can" are aloccurs in the width of the container, which must be increased or decreased according to the number of plates in the element. It will be sufficient, therefore, to describe the A-4 cell. There are also manufactured two cells with smaller plates, for ignition and other light work, the B-2 and B-4 cells, rated

at 40 and 80 ampere-hours respectively. The table below gives the dimensions and the output of the three sizes of vehicle cells:

	A-4,	A-6.	A-8.
Outside measurement of sean, inches	$\begin{array}{c} 2_{13}^{2} \times \\ 5_{18}^{2} \times 12_{8}^{2} \\ 13_{8}^{3} \\ 13.34 \\ 14.21 \\ 30 \\ 1.2 \\ 150 \\ 180 \\ \end{array}$	$\begin{array}{c} 3\frac{1}{4} \times \\ 5\frac{1}{8} \times 12\frac{5}{8} \times 12\frac{5}{8} \\ 19.00 \\ 20.09 \\ 45 \\ 1.2 \\ 225 \\ 270 \end{array}$	$\begin{array}{c} 4\frac{7}{8} \times \\ 5\frac{1}{8} \times 12\frac{3}{8} \\ 13\frac{3}{8} \\ 25.00 \\ 26.15 \\ 60 \\ 1.2 \\ 300 \\ 360 \\ \end{array}$
	j.	I	1

In accordance with the low weight of combination described, a very light but strong wood battery trap is employed, of skeleton construction and with bentwood instead of dovetailed corners. An assembled vehicle battery is claimed to weigh not much more than half as much as a lead battery of the same outputa 64-cell battery for a one-ton truck weighing 900 pounds, as compared with 1,550 pounds in a 44-cell lead battery of the same capacity.

The A-4 Edison cell consists of the following parts: The can, a sheet steel receptacle containing the element and electrolyte, and having a top fitted with stuffing boxes for the positive and negative terminal posts, and with a gas valve and filling aperture; the element, consisting of a group of positive plates containing the nickel active material and connected in multiple with the positive terminal by a nickel-plated steel bolt and nuts, the group being intermeshed with a group of negative plates containing the iron active material and similarly connected in multiple and with the negative terminal; and the electrolyte, consisting of a 21 per cent solution of pure potassium hydrate (caustic potash) in

The metal of the cell is all nickelplated iron, so that a pure nickel surface is everywhere presented, protecting the metal from rust or corrosion. The only other visible materal is the hard rubber of the stuffing boxes and the soft-rubber top of the filling aperture. The can is of sheet steel, with corrugated walls to give the greatest stiffness with minimum weight. The seams are welded by the autogenous method, insuring safety from leakage that might otherwise develop from the severe mechanical vibration and shocks which a vehicle battery receives. The can is embossed in places to receive hard-rubber buttons, to maintain a slight separation or air-gap bebattery tray containing the set of cells.

To the automobile owner the vehicle battery certainly looks good. A compact set of metal cans, with no chance of slopping or breaking. The battery has more the appearance of road service. and smacks less of the electrical laboratory, and this highly mechanical look is justified by the internal construction of the element as a working electrochemical machine. The positive plates, containing the positive or nickel-hydrate active material, consist of a grid of nickeldouble rectangle in shape, with a center of fifteen each, set closely together. The unnecessary to add anything but distilled

tubes are of very thin sheet steel, perplates, therefore, the height and breadth forated with small holes, and packed or "loaded" (very tightly by special maways the same, and the only variation | chines) with nickel hydrate, interspersed with pure nickel in the form of flakes of extreme thinness. The tubes are formed with a double-lapped spiral seam, and are reinforced by rings or ferrules which maintain internal contact between the nickel hydrate and flake nickel and the internal walls of the tube by preventing bulging of the tube by expansion of its contents. This construction is one of the radical improvements over the old Edison cell. The problem was to bring every part of the hydrate .n the plate into active service by insuring permanent electrical contact with its connecting grid, and this result is attained by the thin disks of nickel stacked in the tube in alternation with the layers of hydrate, forming the tube into a mass of intimate physical and electrical contacts between hydrate and nickel and between the edges of the nickel disks and perforated wall of the tube; at the same time permitting intimate contact with the electrolyte in which the tube is immersed, and which seeps in through the perforations. Each of the 41/8-inch tubes in the grid contains about 350 of the tiny nickel disks and their alternating the cell, obtained by the special voltaic layers of hydrate, so that the maximum distance through which the current has to penetrate, in the charge and discharge of the cell, is only one-half the thickness of the exceedingly thin layer of hydrate.

The thin nickel flake is formed by an ingenious machine which deposits alternate electro-platings of copper and nickel, the copper being dissolved out subsequently. So delicate are the nickel disks cut from this electro-plated sheet that a handful of them is as light as so much feathers.

The negative or iron-oxide plates, of the same length and width as the positive plates, consist of a grid of three rows of eight flat, rectangular pockets, each perforated like the tubes of the positive plate with fine holes to permit contact of the electrolyte with the active material. The latter consists of an iron oxide very similar to ordinary "iron rust," subjected to heavy pressure in the containing pockets, so that each mass of active material is practically integral with the grid.

The plates of each group, positive and negative, are hung on their horizontal connecting rods, with spacing bushings between the lugs of the plates and a nut at each end of the rod. Rising from the middle of the rod is the vertical terminal post, having a taper top for making good electrical contact with the adjoining cell or with the external circuit.

The outer surfaces of the two end plates of the element are securely insulated from the walls of the can by sheets of perforated hard rubber. Rods and cross-pieces of the same material serve respectively to separate adjacent plates of the element and to support the element as a whole above the bottom of the can. It is noteworthy that the element does not have to be supported at a considerable distance above the bottom of the cell, as in the lead storage cell, to prevent short circuiting of the tween adjacent cans, and to anchor the plates by the falling of the active macan to its proper place on the wood terial to the bottom of the jar. In the Edison cell there is little or no loss of active material.

The remaining part of the cell, the top or cover of the can, is welded in place by the autogenous method after the element is assembled in the can. The terminal-post stuffing boxes, the gas valve which separates entrained electrolyte from the spraying of the gas in charging, and prevents it from escaping, and the filling aperture with its automatic spring cap, are water-tight, so as to prevent the leakage of electrolyte in service. The plated steel, holding 30 little tubes packed gas valve, by preventing loss of elecwith the active material. The grid is a trolyte during charge, eliminates fuming of the cells, minimizes the amount of rib, and supports the tubes in two rows filling up for evaporation and renders it

## Re-sales tell the story of superiority of

# The New Edison Storage Battery

Whether you are a user or a prospective user of electric vehicles, either pleasure or commercial, read through this chain of evidence:

The Adams Express Company, seven years ago replaced lead battery equipment in four of their delivery trucks, with Edison equipment. Since that time they have purchased over 150 cars, all equipped with Edison Batteries.

Tiffany & Company replaced lead batteries with Edison Batteries in three of their delivery wagons, seven years ago. They have since bought eighteen more cars, all Edison equipped. The Batteries operating these cars have all been in service over a period of from four to seven years.

Hearn & Son, after operating fifteen cars equipped with Edison Battery for three years, discarded lead battery equipment in twelve cars and installed Edison Batteries, making a total of twenty-seven Edison equipped cars.

Frederick Loeser & Company, of Brooklyn, after operating an Edison Battery for one year in one of their vehicles that was made for lead battery equipment, replaced lead batteries with Edison Batteries in six

The United States Express Company have been operating twenty-five vehicles with Edison Battery for several years.

Abraham & Straus, when they decided to put in motor vehicle equipment, purchased twelve vehicles equipped with the

A. A. Vantine & Son have been operating thirteen Edison equipped vehicles over a period of four and a half years.

R. H. Macy & Company have been operating seventeen Edison Battery equipped cars for the past three and a half years.

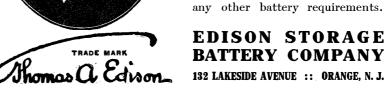
Four of the foremost electric pleasure vehicles are now regularly equipped with the Edison Storage Battery-Detroit, Bailey, Baker, and Waverly.

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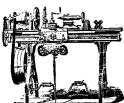
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water during long periods in the life of the cell.

Connections between the cells of a battery and with the external circuit are made by means of heavy nickel-plated copper-wire connectors, with steel lugs swaged on the ends and fitting on the taper terminal posts.

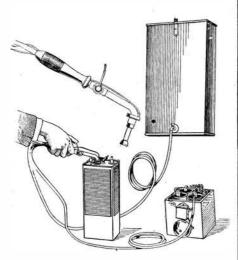
The element is made very compact, with no provision for circulation of electrolyte because the latter is constant, chemically, electrically and physically. In this "can of electricity" the solution is a mere liquid carrier of ions—it has, so to speak, no moods and tenses of its own to be catered to by the battery user. Whereas in the lead cell the constitution (proportions of acid and water), conductivity and specific gravity all undergo definite change during the cycle of charge and discharge, in the Edison cell these qualities remain unaltered. On account of the low voltage of the cell, a vehicle battery of given voltage must contain a larger number of cells than a lead battery, twenty-four lead cells being equivalent to forty of the nickel-iron cells of same ampere-hour capacity; but the low voltage and comparatively high internal resistance are disadvantageous from the purely electrical rather than the vehicle point of view. The cost of charging current, and hence also the efficiency of its recovery in the discharge circuit, is a minor part of the total expense of operating the vehicle; and the low electrical efficiency of the battery is offset by its extremely light weightwhich not only lightens the construction of the vehicle, but increases the proportion of "paying load" carried by the same -and by its ruggedness in service. In estimating the true commercial efficiency of a road-vehicle transportation, the tonmiles obtained must be compared with the total cost of operation of the vehicle, with due allowance for all-around reliability. The nature of the support materials and the method of applying the active material to the plates enable the cell to stand dead short circuits without injury. It is stated that the cell improves in use, instead of deteriorating, and the company guarantees to renew any battery at any time for half the cost of a new battery.

In charging, the rate can be varied through a wide range, and may be far higher than would be safe with the lead battery. "Boosting charges"—given a vehicle battery when partially discharged, so as to carry it through a longer period than its prescribed day's work—may be given for an hour's run at a rate two or three times the normal charging rate: and a regular charge may be completed in a hurry by starting at a high rate and gradually decreasing the rate. A vehicle manufacturer using the battery reports that he charged an exhausted battery of A-4 cells at the rate of 100 amperes for a short time, and states that it pays better to put in over

100 volts when charged can be thrown directly on the 110-volt mains without resistance in series.

dication of its road ability. An electric pleasure vehicle equipped with the batpounds, has climbed Fort George Hill, average speed of 12.32 miles per hour. The cost of the charge was \$1.42, which comes to about 21 cents per day.

The garage work other than charging is simplified by the absence of corrosive



Filling up an Edison cell for evaporation. Fitted with thumb valve and connections to bell circuit.

fumes, the substitution of bolt-and-nut connections for lead-burning, the good retention of the active material by the plates and the absence of "sulphating" and similar troubles, and the non-fragile nature of the cans and elements. It is claimed that a single garage man can readily take care of a very large number of batteries; and in fact the Edison battery is not regarded as very attractive business by the public garages ac- road maps and guide books is that one customed to doctoring lead cells which trolyte and assembled in trays before leaving the factory, and the sealed top to be meddled with by the user, and re-trouble. duces his work, practically, to mere supervision of the charging and filling up and if this route is more than 100 miles for evaporation, an external cleaning long but less than 200 miles, the card is about twice a year (taking the cells reversed and the directions on the back out of their trays and removing dirt are followed. The manufacturers of the from between and underneath the instrument supply cards covering all cans, and removing rust, if any), and known touring routes in the United putting in new electrolyte at long in-States, including several transcontinentervals. The cell is able to stand idle tal routes, and also principal touring for long periods, wholly or partially dis- routes in Europe. About 600 routes are charged, without injury because the catalogued for this country alone. The non-corrosive material of its plates is cards can be ordered by number es immersed in the alkaline electrolyte, the cell being, electrochemically, in a | board case furnished by the manufacstate of stable equilibrium.

The element and its separators fill a large proportion of the cubic contents is small, giving rise to a greater tendency to heat suddenly by excessive cur-

nine-tenths of the normal capacity in a rent than in lead cells. Continuous workshort charge at a high rate than to give ing at high temperatures has an injurithe long, slow charge traditionally ous effect tending to shorten the life of deemed advisable for lead cells. A com- the cell, but a temperature of 100 demercial vehicle battery of high capacity grees is not critical. In hot weather an will give the best results with a 75 to electric fan may be turned on the bat-80 per cent charge at high rate every tery when charging, to keep down its day, thereby avoiding the loss of ef- temperature. The metal cans and the ficiency of charging that comes at the small amount of solution invite more end of a "normal" charge at the pre- rapid chilling in a vehicle standing idle scribed rate. A battery showing about in winter than the heat-insulating hard rubber jars of the lead battery, and is to be avoided by properly inclosing the battery trays, sealing up all holes or open-It is the discharge performance of the ings to avoid the entrance of the cold battery, of course, expressed in "miles air. In charging, however, the battery in the battery," that the auto driver is compartment must be left wide open. most interested in. Hill climbing is a The small amount of electrolyte and the favorite test of an automobile power | liability of heating on charge at the high plant, and the performance of the Edi- charging rates which are permissible, son battery in this respect is a good in- give rise to another sensitive point in the handling of this type of cell, namely, "low electrolyte," i. e., exposing the tops tery, the total weight being 2,387 of the cells, and causing loss of storage capacity. In active service it is necessary a distance of 2,138 feet with 11 per cent to "fill up for evaporation" every few grade, twenty-one times on a single days, using distilled water only (ordincharge. Another interesting performance ary water, which is more or less impure of the battery in city work, on a single or which has been aerated by absorption 71/2-hour charge, is a "run-about" of an of carbonic acid from exposure to the hour and a half to two hours every day air, being injurious to the cell), and putfor seven days. The total weight of the ting in the water just before giving car and two passengers was 2,470 pounds. a charge, to avoid the false level due to and a total of 120 miles was made at an gas bubbles in the electrolyte. Since the closed and sealed top of the cell prevents a view of its contents, a special filler is provided by means of which water may be added to the same height in a large number of cells in rapid succession. This filler is connected by a flexible rubber tube to an overhead covered tank of distilled water, and is also wired to an electric bell and dry battery. The filler spout is inserted in the aperture of the cell, and a thumb-valve on the filler is operated to allow the water to flow into the can. The stream of distilled water presents a high resistance in the bell circuit, but when the level of the electrolyte has risen into contact with the end of the spout, the bell rings. It is important to avoid fill ing the cans too full, for this leaves no room for the gas to escape during charge and the electrolyte is liable to be forced out in bubbles-a cause of low

The potash solution becomes contaminated in time, from impurities that have accidentally gotten into the cell either directly or by way of the filler spout (which may be laid down carelessly on a dirty bench), or by absorption of gases in the garage. It therefore needs to be replaced by fresh electrolyte about once in eight months for a commercial-vehicle battery or once a year for a pleasurevehicle battery.

## **Some Mechanical Road Guides**

(Continued from page 33.)

must first know his precise location on have been abused, or worn out, in vehicle the road before he can tell when and service. The cells are filled with elec- how to proceed; a mistake of even half a mile in the distance from a given point then making a turn might throw one means that the element is not expected entirely off the course and cause much

One card serves for only one route, wanted, and kept in good order in a card-

Should the tourist wish to make a detour from the regular route or take a of the can, and the amount of electrolyte side excursion, the instrument can be reset by means of the screw in the back to bring the card to correspond with



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